

Research and Practical Insights into Using Gypsum

# The Effect of Gypsum on Soil Physical Properties Dr. Jerry Bigham

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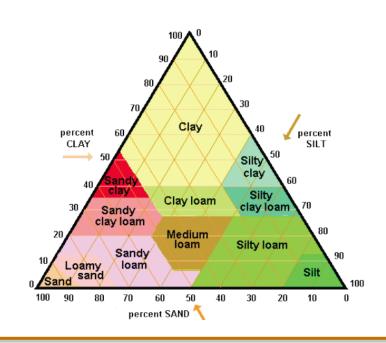
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### Soil Physical Properties

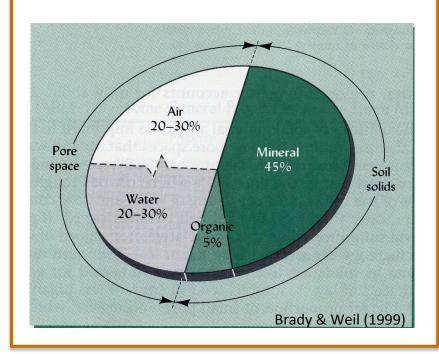
# Some are almost invariant

- Particle Size Distribution (Texture)
- Mineralogy

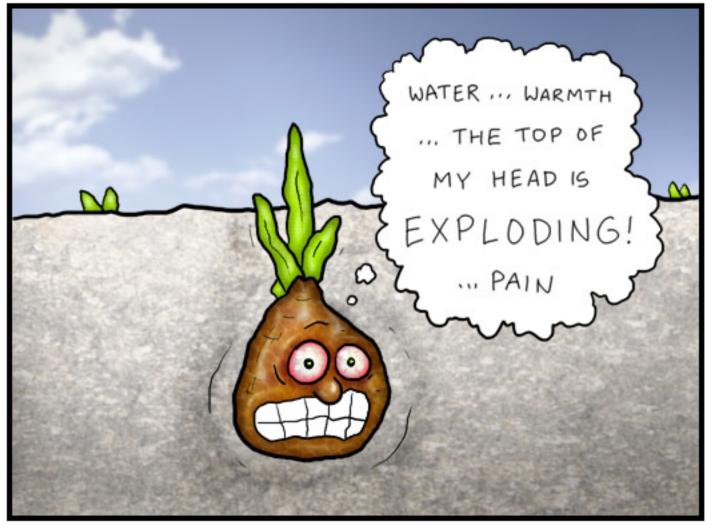


#### Others are sensitive to mgt.

- Structure (Type & Size)
- Porosity (Amount, Size, Continuity)



#### **DOCTOR FUN**



David Farley, d-farley@ibiblio.org Copyright © 2001

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What springtime is really like for bulbs

# Air-water balance is the single most important factor limiting agricultural production in the U.S.

According to a study by Mittler (2006), the top two causes of economic loss to U.S. agriculture between 1980 and 2004 (major events of \$1B loss or more) were:

- 1. Combined heat and drought stress (\$130B)
- 2. Flooding and water-logging (\$50B)





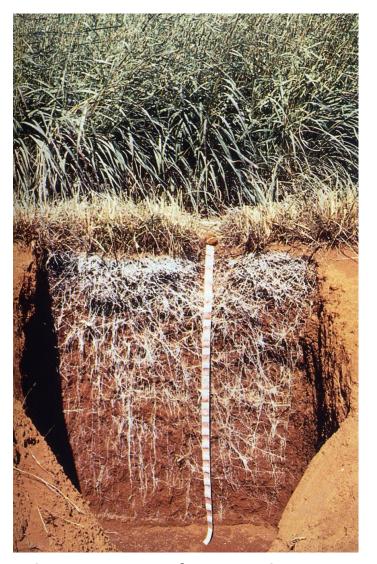
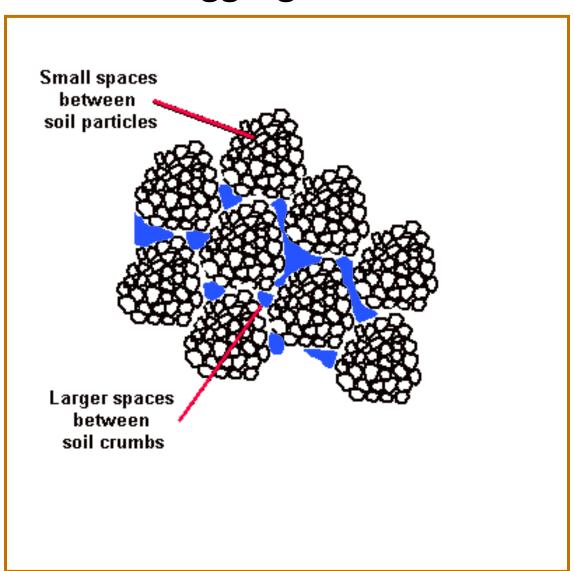


Photo courtesy of S.W. Buol

The ultimate goal of air & water management is to create an environment where the plant root system can reach its full genetic potential for respiration, nutrient uptake, and water transmission. Usually, this means exploiting the largest possible soil volume.

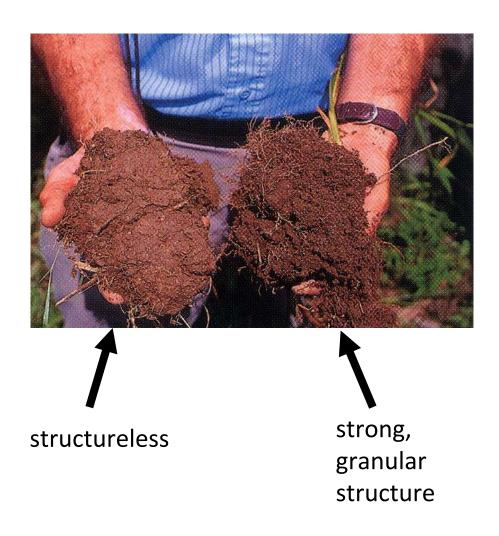
## Several dynamic soil properties are impacted by the creation of water stable aggregates

- Infiltration
- Percolation (K<sub>sat</sub>)
- Available water
- Gas exchange



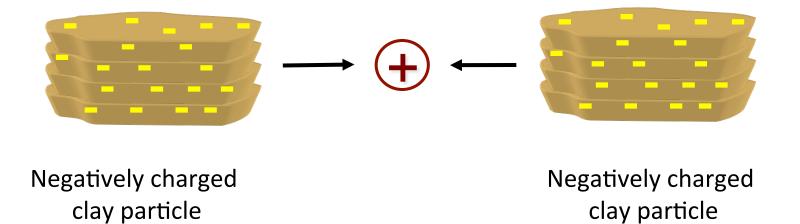
# Aggregation and soil structure are enhanced by:

- root & faunal activity
- soil organic matter
- microbial exudates
- soluble calcium ions

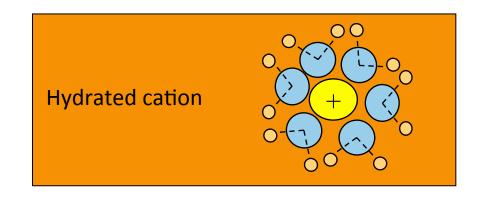


Calcium (Ca<sup>2+</sup>) stabilizes decayed soil organic matter (humus) and is one of several cations (positively charged ions or molecules) that may occupy sites on the soil CEC. Others include ammonium (NH<sub>4</sub><sup>+</sup>) sodium (Na<sup>+</sup>), potassium (K<sup>+</sup>), magnesium (Mg<sup>2+</sup>), and aluminum (Al<sup>3+</sup>).

These cations influence the tendency of soil colloids (clay particles) to separate (disperse) or aggregate (flocculate).



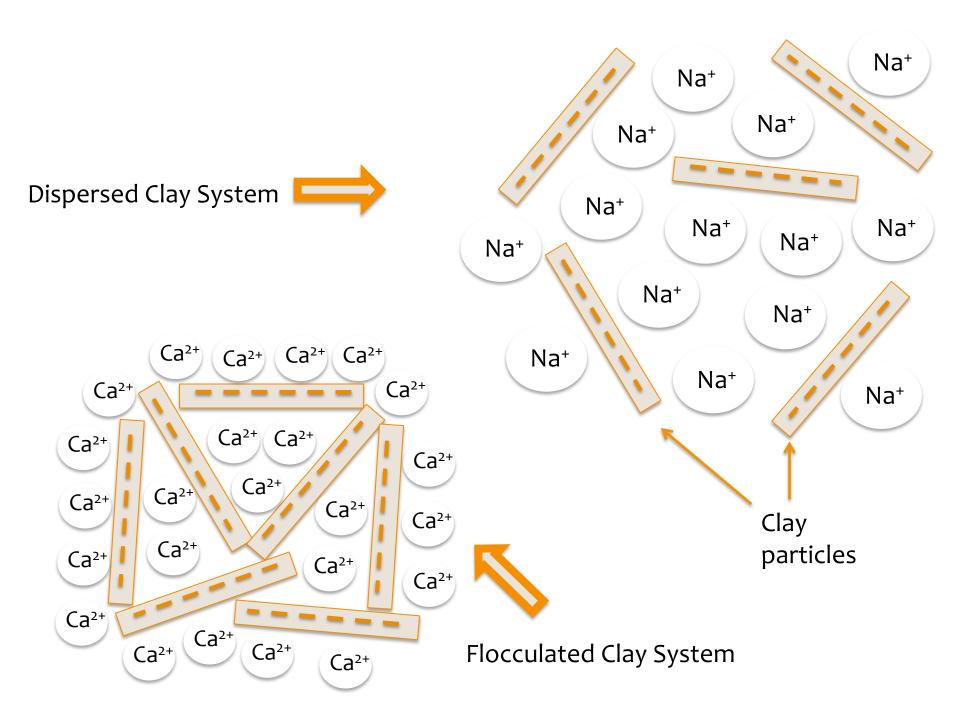
The tendency of a cation to serve as a dispersant or flocculant depends mostly on its charge and hydrated radius within the soil system.



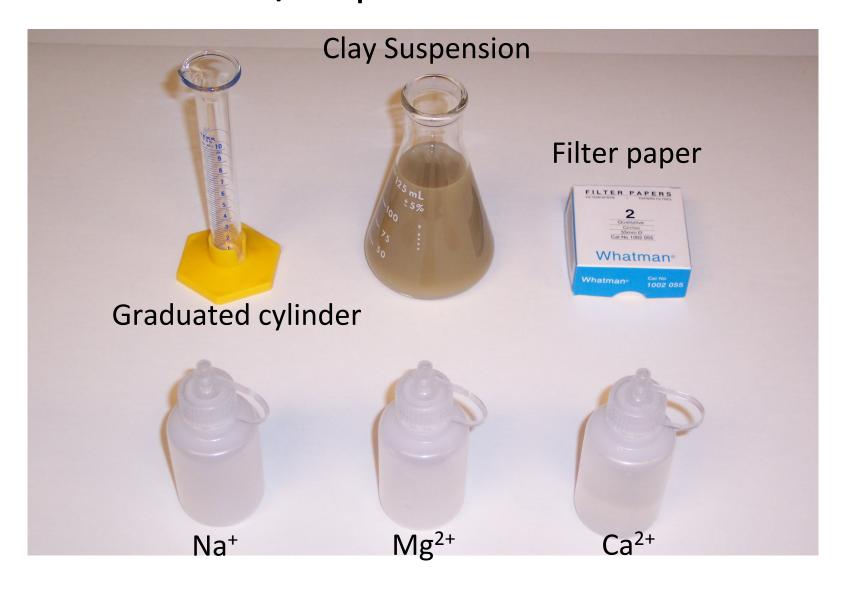
Cation	Charge per ion	Hydrated radius(Å) per unit charge <sup>*</sup>	Relative flocculating power**
Sodium (Na)	1	3.6	1.0
Magnesium (Mg)	2	2.2	27.0
Calcium (Ca)	2	2.0	43.0

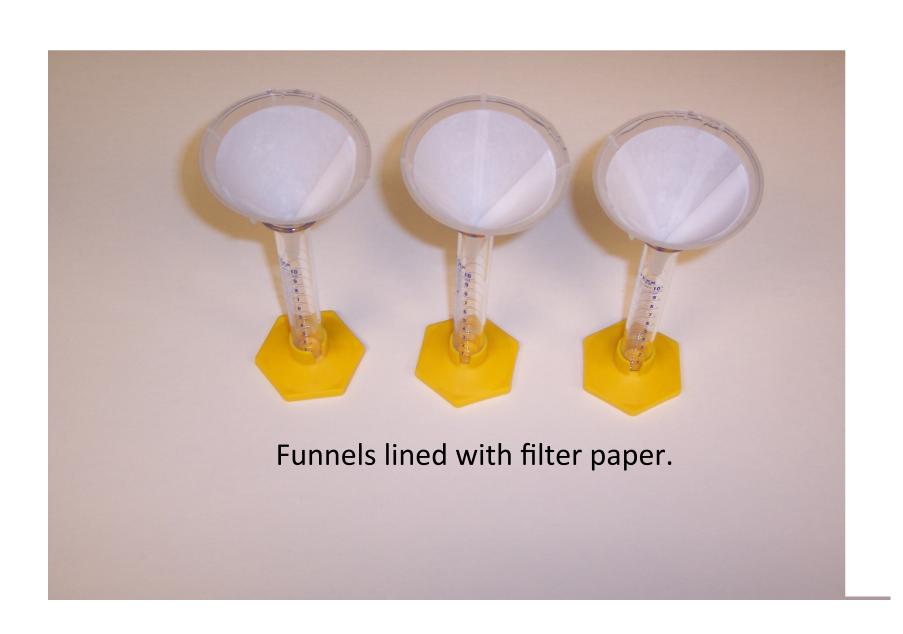
<sup>\*</sup>Conway (1981)

<sup>\*\*</sup> Rengasamy & Sumner (1998)

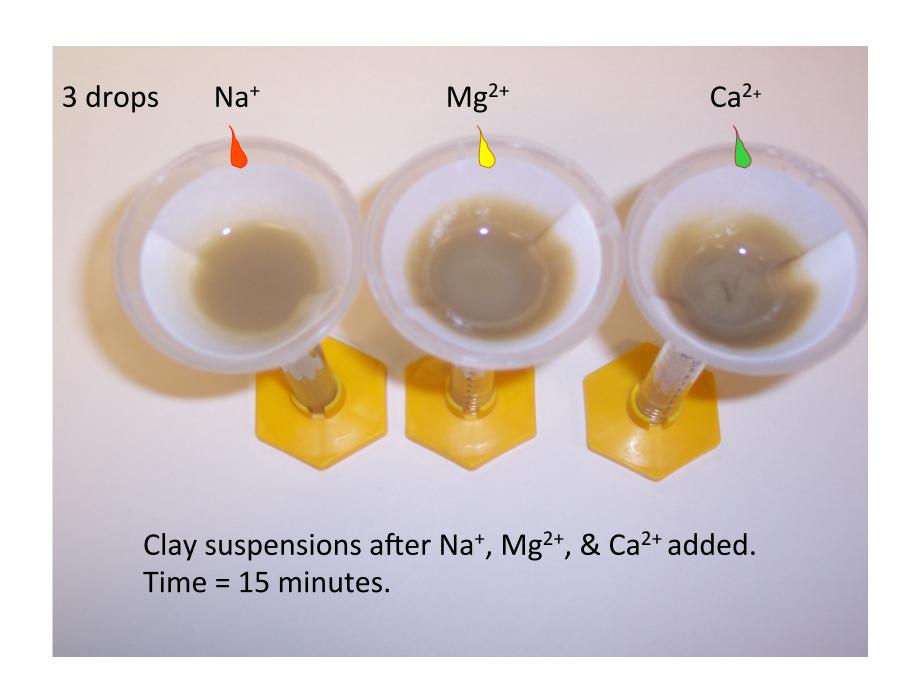


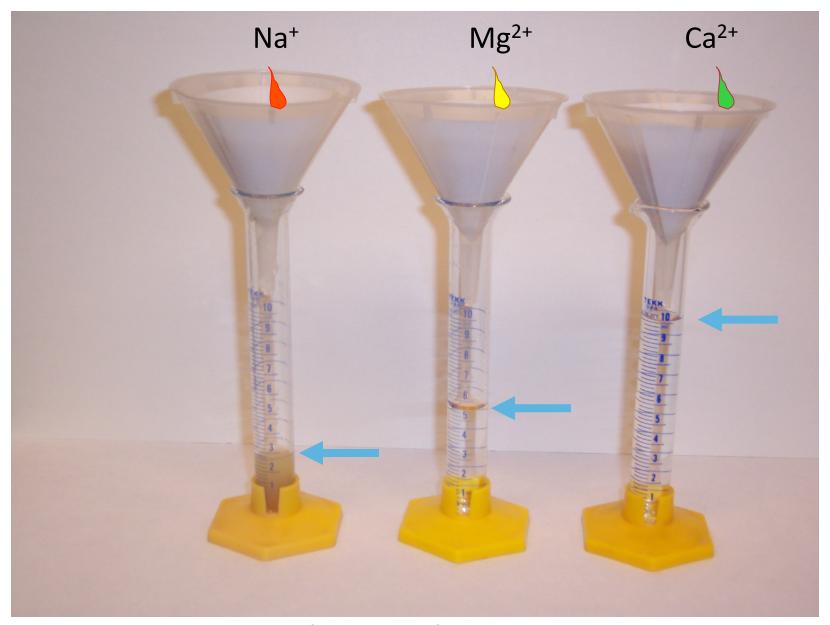
### Flocculation/Dispersion Demonstration











Filtrates after  $Na^+$ ,  $Mg^{2+}$ , and  $Ca^{2+}$  added. Time = 15 minutes.

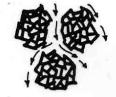


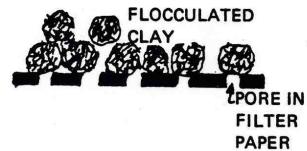
### Na



What is happening?

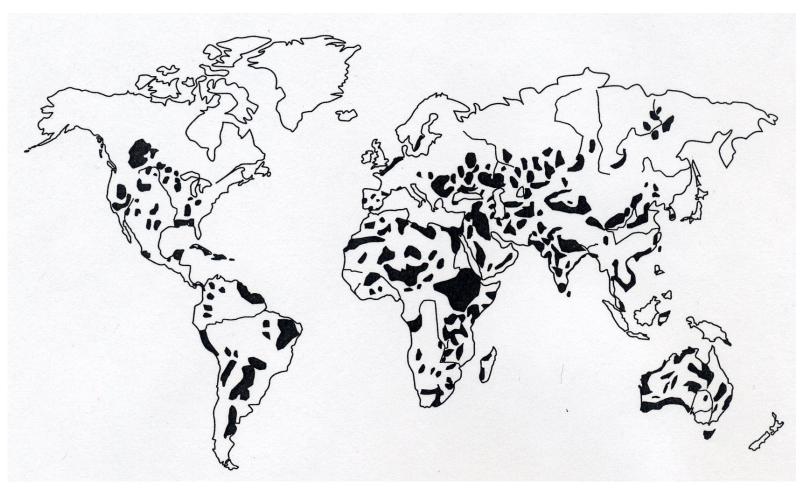




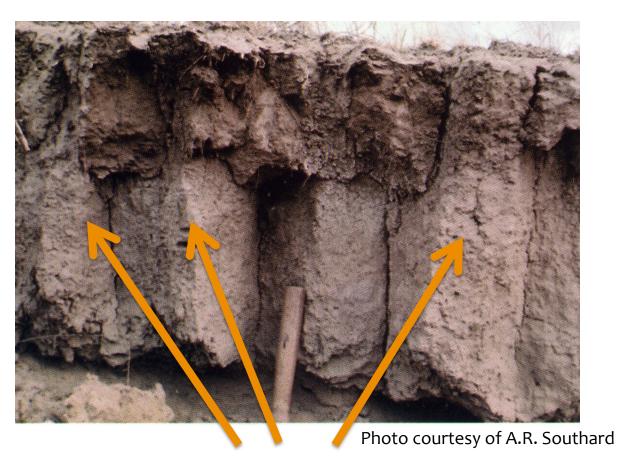


#### Global Distribution of Na-affected Soils

(from Sparks, Environmental Soil Chemistry, 2<sup>nd</sup> Ed (2003)



Sodic (sodium affected) soils: have exchangeable Na <u>></u> 15 % of the CEC, <u>and</u> exchangeable Na + Mg exceeds exchangeable Ca (USDA Hndbk 436, 1st ed, 1976).



Note the large, internally massive prisms occurring in the subsoil.



Deep tillage in southern CA to improve permeability of a sodic (high sodium) soil.



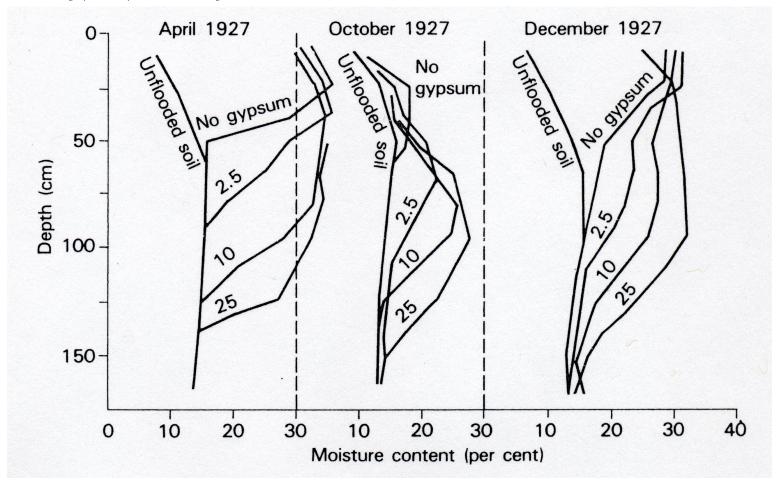
Photos courtesy of Blake Sanden

### Gypsum as a Soil "Conditioner"

Gypsum has been used for many years to improve aggregation (structure) and inhibit or overcome dispersion in sodic (high sodium) soils. Soil dispersion contributes to:

- surface sealing/crusting
- problems with seedling emergence
- runoff/erosion
- subsoil swelling with decreased water and root penetration
- poor air exchange

Reclamation of an impermeable soil with 10% exchangeable Na using surface applied gypsum at rates of 2.5, 10, and 25 ton/ha

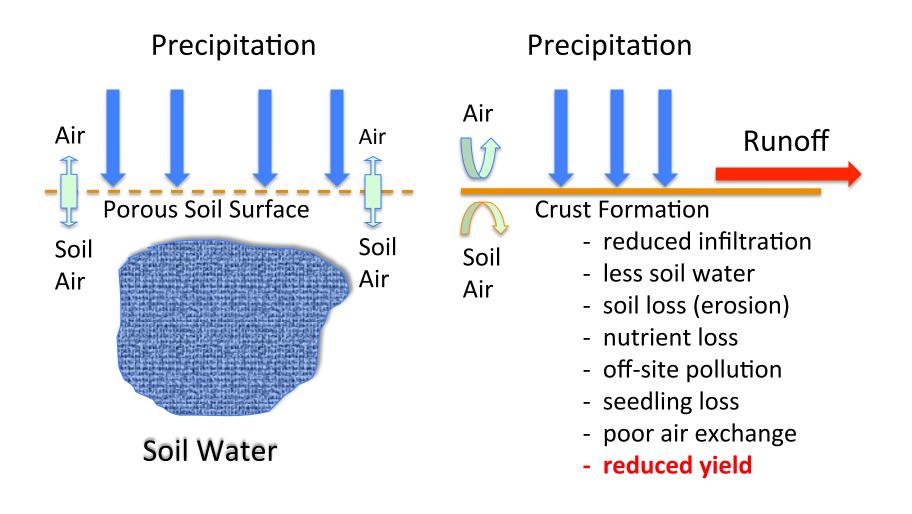


Source: Greene, H.J. J. Agric. Sci. 18:531 (1928)

What about non-sodic soils in the corn belt? Is there really a place for gypsum in a comprehensive management plan?



Clay dispersion and collapse of structure at the soilair interface is a major contributor to surface sealing in **both** sodic and non-sodic soils.





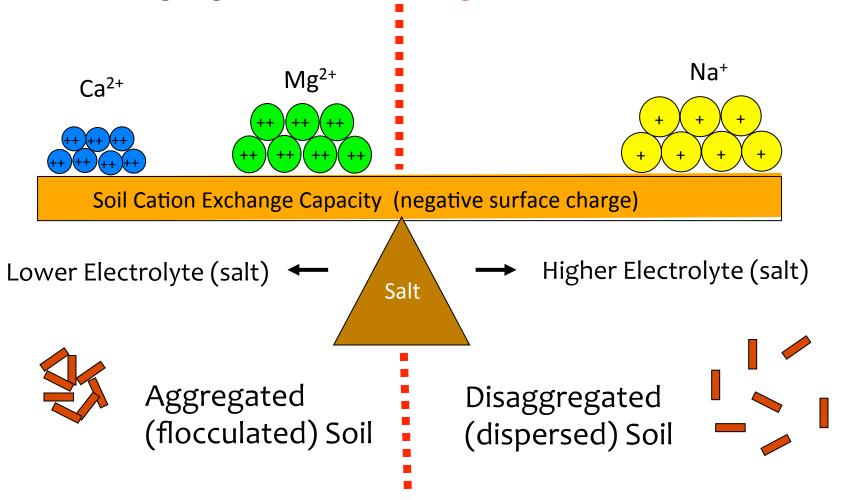
Natural rainwater is erosive, in part, because of its low electrolyte (salt) content.

Work at the NSEL (Norton et al.) has shown that soil and chemical loss due to crusting, poor infiltration of rainfall, and runoff can be reduced by managing the calcium (Ca) status of the topsoil.

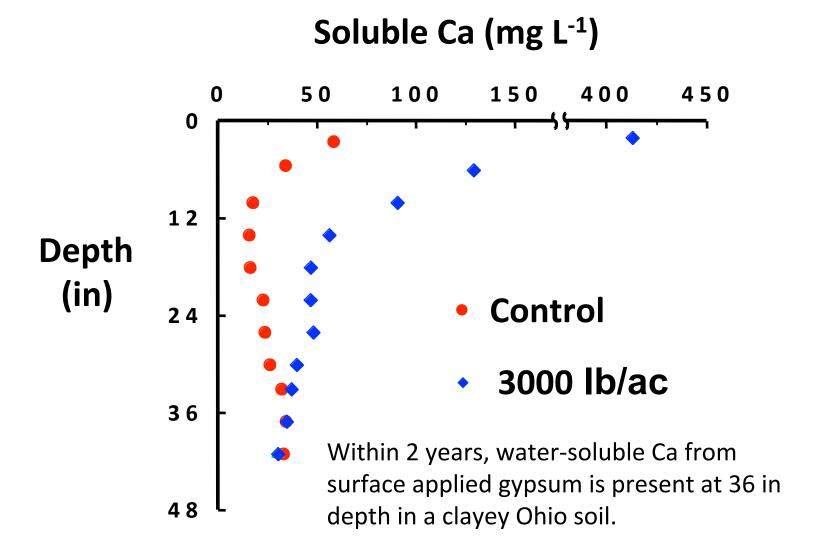
Traditional sources of Ca are:

<u>Compound</u>	<u>Solubility</u>
Limestone (CaCO <sub>3</sub> )	0.014 g/L
Dolomite [CaMg( $CO_3$ ) <sub>2</sub> ]	0.32 g/L
Gypsum (CaSO <sub>4</sub> )	2.41 g/L

Aggregate stability largely depends on the balance between Exch. Ca<sup>2+</sup>, Mg<sup>2+</sup> and Na<sup>+</sup> as well as the amount of total electrolyte (salt) in the soil soln. Exchangeable Ca<sup>2+</sup> is a good aggregating (flocculating) agent; Na<sup>+</sup> is not; Mg<sup>2+</sup> is intermediate.



Because of its solubility, gypsum can also have a significant, positive impact on subsoil chemistry and structure, even under rainfed agriculture.

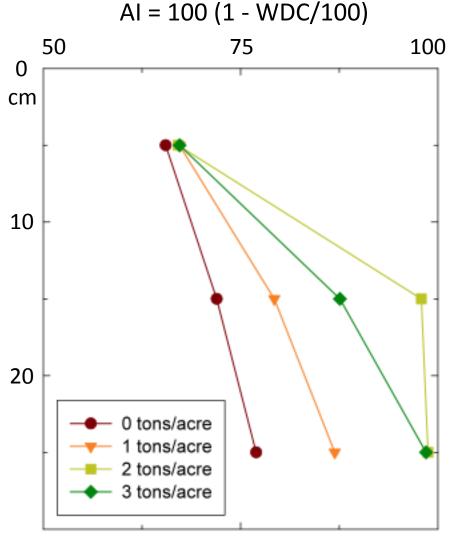


Effects of surface-applied gypsum on aggregation index (AI) in a silty, no-till cotton soil after three growing seasons.



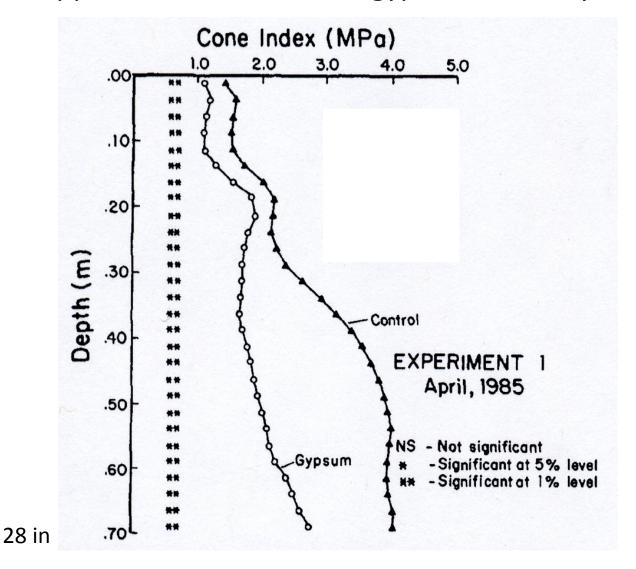
Grenada soil

Rhoton & McChesney (2011)



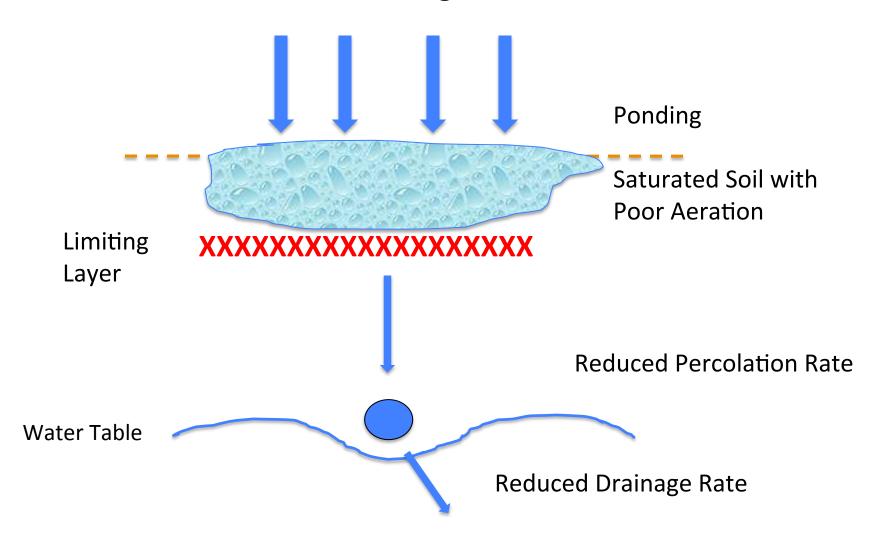
Depth

Mechanical impedance (root resistance) with depth, 3 yr after surface application of 3200 lb/ac gypsum on a clay soil

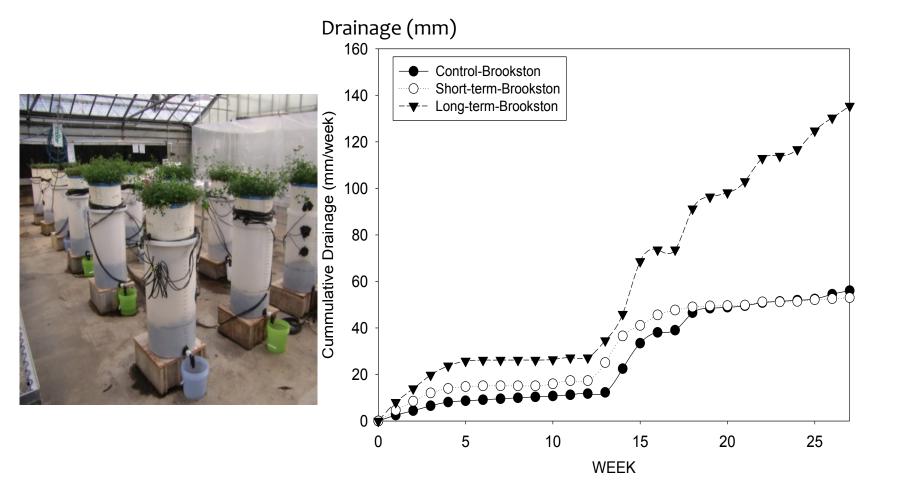


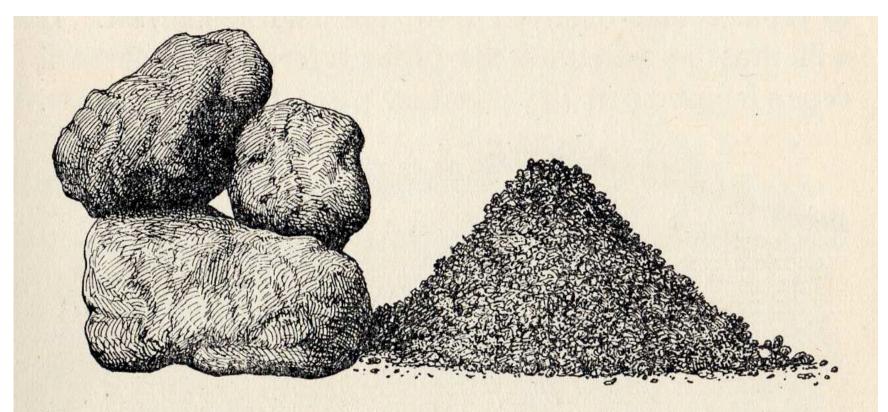
Radcliffe et al. (1986)

#### Rainfall or Irrigation



Cumulative drainage over a 25 week period from intact cores (30.5 x 75 cm) of Brookston soil after 0, 4, and 12 yr of surface applied gypsum at a rate of 1,500 lb/yr (Tirado-Corbala, 2010).





The physical condition of the soil is as important as its chemical composition.

The lumpy soil contains as much plant food as the friable soil, but the plant roots cannot penetrate the hard lumps to obtain it

Alfred Vivian. 1912. "First Principles of Soil Fertility"